

SYNTHESIS OF BRIDGED RING SYSTEMS RELATED TO CLOVENE\*

N.G. Kundu\*\*, S.L. Mukherjee and P.C. Dutta

Department of Organic Chemistry

Indian Association for the Cultivation of Science, Jadavpur, Calcutta-32

(Received 27 March 1962; in revised form 16 May 1962)

IN view of a recent publication by Murray *et al.*<sup>1</sup>, we are prompted to report our results on a synthetic approach to the ring-system related to clovene. The novelty of the present communication is the synthesis of a hydrindane derivative (XII) with appropriate functional groups and having the desired stereochemistry at all the asymmetric centres.

The bicyclic ketone V was singled out to be a potential intermediate for this purpose in view of the successful observations made in this laboratory in a related field and the synthesis of the ketone was achieved through the following steps. The conjugate addition of methylmagnesium iodide to ethyl  $\alpha$ -cyano- $\beta$ -(*p*-methoxyphenyl)-crotonate I<sup>2</sup> was effected in the presence of cuprous iodide<sup>3</sup> resulting in the formation of the product II, b.p. 167-174°/0.6 mm.  $\beta$ -Methyl- $\beta$ -(*p*-methoxyphenyl)-butyric acid (III), m.p. 89-90° obtained on hydrolysis and decarboxylation was cyclized with polyphosphoric acid<sup>4</sup> to the indanone IV, red 2,4-dinitrophenylhydrazone (m.p. 266-267°).

---

\* Analysis of all the compounds described corresponded to the calculated values.

\*\* Present address: Department of Chemistry, University of Southern California, Los Angeles 7, California, U.S.A.

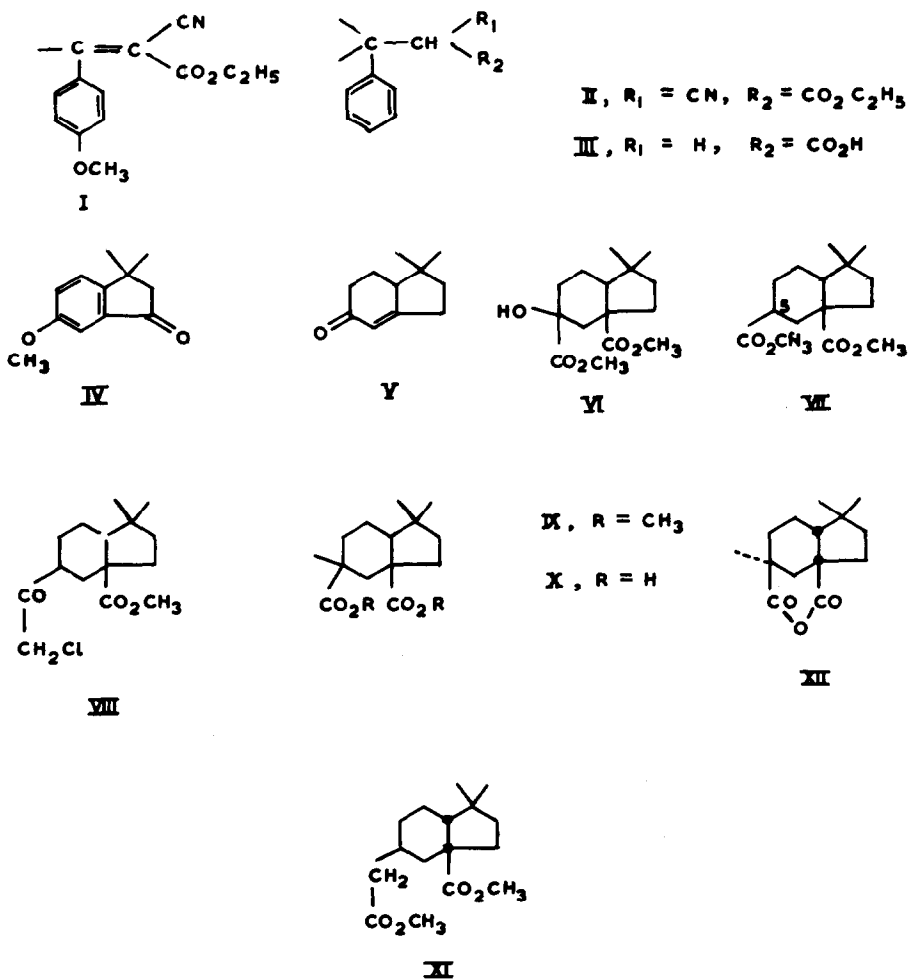
<sup>1</sup> R.D.H. Murray, W. Parker and R.A. Raphael, Tetrahedron **16**, 74 (1961).

<sup>2</sup> W. Baker, J.F.W. McOmie and A.S. Weaving, J. Chem. Soc. 2018 (1956).

<sup>3</sup> A.J. Birch and R. Robinson, J. Chem. Soc. 501 (1943); W. Parker and R.A. Raphael, Ibid. 1723 (1955).

<sup>4</sup> Sukh Dev, J. Indian Chem. Soc. **32**, 255 (1955).

The reduction of the ketone to the corresponding hydrocarbon was successfully effected through catalytic hydrogenation in presence of small amount of perchloric acid.<sup>5</sup> The aromatic ring of the indane was further reduced with sodium in liquid ammonia<sup>6</sup> and the resulting product isomerized with



<sup>5</sup> K.W. Rosenmund and E. Karg, *Ber. Dtsch. Chem. Ges.* **75**, 1850 (1942);  
 W.S. Johnson, R.G. Christiansen and R.E. Ireland, *J. Amer. Chem. Soc.* **79**,  
 1995 (1957).

<sup>6</sup> A.J. Birch, *Quart. Rev.* **4**, 69 (1950).

dilute sulphuric acid. The unsaturated ketone V [ $\lambda_{\max}$  230 m $\mu$  ( $\log \epsilon$  3.9)], thus obtained, afforded a red 2,4-dinitrophenylhydrazone (m.p. 168<sup>o</sup>) and a semicarbazone (m.p. 215<sup>o</sup>). An acidic material could be isolated in a satisfactory yield on conjugate addition of potassium cyanide to V in aqueous alcoholic solution and subsequent alkaline hydrolysis and this on esterification with diazomethane afforded a neutral liquid, VI, b.p. 147-155<sup>o</sup>/4 mm. VI was further converted to the chloro-compound and finally to VII on boiling with zinc dust in acetic acid. In order to introduce the methyl group at 5-position in VII this was partially hydrolysed to the ester-acid, characterized through the crystalline anilide (m.p. 121<sup>o</sup>). This was successively converted to the acid-chloride, to diazoketone with diazomethane and finally to the chloro-ketone VIII with dry hydrogen chloride.<sup>7</sup> This was subjected to Favorskii rearrangement in the presence of sodium methoxide<sup>8</sup> and the desired diester IX was isolated in a poor yield, the major product being XI. Drastic alkaline hydrolysis of IX afforded the di-acid X, m.p. 171-172<sup>o</sup> which underwent easy conversion to the anhydride XII, m.p. 98-99<sup>o</sup> ( $\nu_{\max}$  5.56  $\mu$ ; 5.68  $\mu$ ). Development of the third ring, characteristic of the clovene ring-system, is in active progress along well-tried methods.

Acknowledgements - Our thanks are due to Sri Anil Kumar Kundu for his material assistance particularly in the latter part of the work. Our thanks are due to the Chemical Society for a grant from the Chemical Society Research Fund and to C.S.I.R., Government of India for financial assistance to one of us (N.G.K.).

---

<sup>7</sup> Ch.R. Engel and G. Just, J. Amer. Chem. Soc. 76, 4909 (1954).

<sup>8</sup> R.B. Loftfield and L. Schaad, J. Amer. Chem. Soc. 76, 35 (1954); H.O. House and W.F. Gilmore, Ibid. 83, 3980 (1961).